

Program	Master of Pharmacy (M.Pharm)	Semester - 1
Type of Course	-	
Prerequisite		
Course Objective	-	
Effective From A.Y.	2023-24	

Teaching Scheme (Contact Hours)					Exa	mination Sch	eme				
				Т   Т		Theory	Marks	Practical Marks		Total	
Lecture	Tutorial	Lab	Credit	External Marks (T)	Internal Marks (T)	External Marks (P)	Internal Marks (P)	Marks			
-	-	12	6	-	-	100	50	150			

SEE - Semester End Examination, CIA - Continuous Internal Assessment (It consists of Assignments/Seminars/Presentations/MCQ Tests, etc.)

Cou	rse Content	<b>T</b> - Teaching Hours   <b>W</b> - Wei	ghtag
Sr.	Topics	Т	W
1	Practical	20	5
	Analysis of Pharmacopoeial compounds in bulk and in their formulations (tablet/ capsules/ s spectrophotometer	semisolids) by UV Vis	
2	Practical	12	10
	Simultaneous estimation of multi-drug component containing formulations by UV spectropho	tometry	
3	Practical	8	5
	Experiments based on HPLC		•
4	Practical	4	3
	Experiments based on Gas Chromatography		•
5	Practical	8	5
	Estimation of riboflavin/quinine sulphate by fluorimetry		
6	Practical	4	2
	Estimation of sodium/potassium by flame photometry or AAS		
7	Practical	16	5
	Case studies on:		
8	Practical	4	5
	Development of Stability study protocol		•
9	Practical	4	5
	Estimation of process capability		•
10	Practical	30	10
	In process and finished product quality control tests for tablets, capsules, parenterals and se	misolid dosage forms.	1
11	Practical	8	5



Cour	rse Content	<b>T</b> - Teaching Hours   <b>W</b> -	Weig	ghtag
Sr.	Topics		Т	W
	Assay of raw materials as per official monographs			
12	Practical		4	5
	Testing of related and foreign substances in drugs and raw materials	'		
13	Practical		16	5
	To carry out pre formulation study for tablets, parenterals	'		
14	Practical		4	5
	To study the effect of pH on the solubility of drugs	'		
15	Practical		6	5
	Quality control tests for Primary and secondary packaging materials	'		
16	Practical		8	5
	Accelerated stability studies	'		
17	Practical		8	5
	Improved solubility of drugs using surfactant systems	'		
18	Practical		8	5
	Improved solubility of drugs using co-solvency method	'		
19	Practical		8	5
	Determination of Pka and Log p of drug	'		1
	•	Total	18	100

Cuggosted Dietribution	Not Theory Marks He	sing Bloom's Taxonomy
Suggested Distribution	i di Tilediy Maiks Us	illy bloom's laxonolly

Level	Remembrance	Understanding	Application	Analyze	Evaluate	Create
Weightage	20	30	20	10	10	10

NOTE: This specification table shall be treated as a general guideline for the students and the teachers. The actual distribution of marks in the question paper may vary slightly from above table.

## Course Outcomes

Cour	SE OUTCOMES CONTROL OF THE PROPERTY OF THE PRO				
At the	At the end of this course, students will be able to:				
C01	Basic knowlegde of the analytical methods for estimation of drugs in single and combined dosage form				
C02	Ability to perform preformulation study of drugs, in pocess and finished product quality control tests for vaious dosage forms and enhancement of solubity of drugs				

Printed on: 31-12-2024 01:56 PM Page 2 of 4



## **Reference Books**

	ir chiec Books	
1.		nistry (TextBook) n   Wiley India   6TH, Pub. Year 2007
2.	Pharmaceutical By D.G. Watson	<b>Analysis</b>   Harcourt Publishers Ltd   1ST, Pub. Year 2000
3.		armaceutical Analysis (TextBook) s   John Wiley & Sons   3RD, Pub. Year 2004
4.		naceutical chemistry (TextBook) J.B.Stenlake   CBS PUBLISHER   4TH, Pub. Year 2010
5.	Indian Pharmac By Government 2022	of India, Ministry of Health and Family Welfare   Indian Pharmacopoeia Commission, Ghaziabad, India, Pub. Year
6.		ok of Quantitative Chemical Analysis I,RC DENNEY,J D BARES,M THOMAS,B SIVASANKAR   PEARSON   6 TH, Pub. Year 1989
7.		dbook in Instrumental analysis (TextBook) Patel, Dr.Pruvi shah, Hitesh Raval   Nirav Prakashan   1ST, Pub. Year 2013

Printed on: 31-12-2024 01:56 PM Page 3 of 4



## **List of Practical**

LISTO	T Practical					
1.	To carry out as:	say of Paracetamol by UV spectrophotometric method as per the IP 2022				
2.	To determine %	w/w of Paracetamol in given tablet by colorimetric method.				
3.	To determine %w/w of Ofloxacin in given tablet by Colorimetric method.					
4.	To perform assay of Indomethacin using UV spectroscopic method.					
5.	Determination of	of Chloramphenicol from its ointment by UV-Visible Spectroscopy.				
6.	To estimate the	amount of quinine sulphate present in given sample by photo fluorimetry.				
7.	To determine a	mount of Caffeine and Sodium Benzoate in mixture by UV spectroscopy using simultaneous equation method.				
8.	To determine a	mount of Caffeine and Sodium Benzoate in mixture by UV spectroscopy using absorbance ratio method.				
9.	To determine a	mount of Ornidazole and Ofloxacin by Simultaneous equation method.				
10.	To determine a	mount of Ornidazole and Ofloxacin by absorption ratio method.				
11.	To determine a	mount of Ornidazole and Ofloxacin by 1 st order derivative method.				
12.	To determine a	mount of Paracetamol and Ibuprofen by absorption ratio method in Combiflam.				
13.	To determine a	mount of Paracetamol and Ibuprofen by simultaneous equation method in Combiflam.				
14.	To determine a	mount of Paracetamol and Ibuprofen by 1 st order derivative method in Combiflam.				
15.	To estimate Pa	racetamol by HPLC method.				
16.	Demonstration	for Development of stability Study Protocol for API & Finished Pharmaceutical products				
17.	To carry out Pre	eformulation study for tablets (Paracetamol)				
18.	To perform in p	rocess and finished product Quality Control tests for tablet (Paracetamol)				
19.	To carry out Pre	eformulation study for Parentrals (Diclofenac Sodium)				
20.	To perform in p	rocess and finished product quality control test for Parentrals (Diclofenac sodium)				
21.	To perform in p	rocess and finished product quality control tests for semisolid dosage forms				
22.	To carry out Pre	eformulation study for tablets (Aspirin)				
23.	To perform in p	rocess and finished product Quality Control tests for tablet (Aspirin)				
24.	To Perform Ass	say of Raw materials as per official monographs				
25.	To study the ef	fect of pH on the solubility of drugs (Paracetamol)				
26.	To perform Qua	lity control testes for Primary and secondary packaging materials				
27.	To perform Acc	elerated stability studies				
28.	To study the im	proved solubility of drugs (Paracetamol) using surfactant system				
29.	To study the im	proved solubility of drugs (Amlodipine) using co-solvency method				
30.	To determine P	ka and Log P of Salicylic acid				
31.	To study case s	study on Out of Specification (OOS)				
32.	To study case s	study on Out of Trend (OOT)				
33.	To study case s	study on Six Sigma				
34.	To study case s	study on total quality management (TQM)				
35.	To study case s	study on Corrective and Preventive Action (CAPA)				
	•					

Printed on: 31-12-2024 01:56 PM Page 4 of 4